

# Benzenamine, 4,4'-methylenebis[N,N-dimethyl-

## Other names:

Aniline, 4,4'-methylenebis[N,N-dimethyl-  
Arnold's base  
p,p'-Bis(dimethylamino)diphenylmethane  
4,4'-Bis(dimethylamino)diphenylmethane  
Bis[p-(N,N-dimethylamino)phenyl]methane  
Bis[p-(dimethylamino)phenyl]methane  
Bis[4-(dimethylamino)phenyl]methane  
Bis[4-(N,N-dimethylamino)phenyl]methane  
4,4'-Bis(dimethylaminophenyl)methane  
4,4'-Methylenebis[N,N-dimethylaniline]  
Michler's base  
Michler's hydride  
Michler's methane  
Reduced michler's ketone  
Tetrabase  
Tetramethyldiaminodiphenylmethane  
N,N,N',N'-Tetramethyl-p,p'-diaminodiphenylmethane  
N,N,N',N'-Tetramethyl-4,4'-diaminodiphenylmethane  
p,p'-Tetramethyldiaminodiphenylmethane  
4,4'-Tetramethyldiaminodiphenylmethane  
p,p-Dimethylaminodiphenylmethane  
p,p-Tetramethyldiaminodiphenylmethane  
p,p'-Bis(N,N-dimethylaminophenyl)methane  
Diphenylmethane, tetramethyldiamino-  
Methane, bis(p-(dimethylamino)phenyl)-  
Methanediamine, tetramethyl-N,N'-diphenyl-  
N,N'-Tetramethyldiaminodiphenylmethane  
NCI-C01990  
4,4'-(Dimethylamino)diphenylmethane  
BAZE michlerova  
4,4'-Methylenebis(N,N-dimethyl)benzenamine  
4,4'-Methylene bis(N,N'-dimethylaniline)  
Methane base  
Methylene base  
p,p'-Tetramethyldiamindiphenylmethane  
Methane, bis(4,4'-dimethylaminophenyl)-  
NSC 36782  
N,N,N',N'-tetramethyl-4,4'-methylenedianiline

**Inchi:** InChI=1S/C17H22N2/c1-18(2)16-9-5-14(6-10-16)13-15-7-11-17(12-8-15)19(3)4/h5-12H,1

**InchiKey:** JNRLEMMIVRBKJE-UHFFFAOYSA-N

**Formula:** C17H22N2  
**SMILES:** CN(C)c1ccc(Cc2ccc(N(C)C)cc2)cc1  
**Mol. weight [g/mol]:** 254.37  
**CAS:** 101-61-1

## Physical Properties

Property code	Value	Unit	Source
gf	519.38	kJ/mol	Joback Method
hf	190.97	kJ/mol	Joback Method
hfus	33.13	kJ/mol	Joback Method
hvap	63.40	kJ/mol	Joback Method
ie	6.72	eV	NIST Webbook
ie	7.10	eV	NIST Webbook
log10ws	-3.39		Crippen Method
logp	3.409		Crippen Method
mcvol	222.830	ml/mol	McGowan Method
pc	2025.41	kPa	Joback Method
tb	663.00	K	NIST Webbook
tc	897.29	K	Joback Method
tf	424.17	K	Joback Method
vc	0.807	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	602.52	J/mol×K	676.56	Joback Method
cpg	621.33	J/mol×K	713.35	Joback Method
cpg	638.83	J/mol×K	750.14	Joback Method
cpg	655.10	J/mol×K	786.92	Joback Method
cpg	670.23	J/mol×K	823.71	Joback Method
cpg	684.29	J/mol×K	860.50	Joback Method
cpg	697.36	J/mol×K	897.29	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	456.50 ± 1.50	K	0.40	NIST Webbook

## Sources

Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C101611&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C101611&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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