

# Cholest-7-en-3-ol, acetate, (3«beta»)-

<b>Other names:</b>	Cholest-7-en-3«beta»-ol, acetate 3«beta»-Acetoxycholest-7-ene Cholest-7-en-3-ol acetate 7-Cholestenol acetate
<b>Inchi:</b>	InChI=1S/C29H48O2/c1-19(2)8-7-9-20(3)25-12-13-26-24-11-10-22-18-23(31-21(4)30)14
<b>InchiKey:</b>	BHZSLZOJNSDNOI-JBJFYEOLSA-N
<b>Formula:</b>	C29H48O2
<b>SMILES:</b>	CC(=O)OC1CCC2(C)C(CC=C3C2CCC2(C)C3CCC2C(C)CCCC(C)C)C1
<b>Mol. weight [g/mol]:</b>	428.69
<b>CAS:</b>	17137-70-1

## Physical Properties

Property code	Value	Unit	Source
gf	123.22	kJ/mol	Joback Method
hf	-621.08	kJ/mol	Joback Method
hfus	40.10	kJ/mol	Joback Method
hvap	86.76	kJ/mol	Joback Method
log10ws	-8.44		Crippen Method
logp	7.960		Crippen Method
mcvol	379.170	ml/mol	McGowan Method
pc	927.81	kPa	Joback Method
rinpol	3228.00		NIST Webbook
rinpol	3228.00		NIST Webbook
tb	977.25	K	Joback Method
tc	1206.29	K	Joback Method
tf	561.27	K	Joback Method
vc	1.438	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1432.94	J/mol×K	977.25	Joback Method
cpg	1465.33	J/mol×K	1015.42	Joback Method
cpg	1498.23	J/mol×K	1053.60	Joback Method

cpg	1531.96	J/mol×K	1091.77	Joback Method
cpg	1566.85	J/mol×K	1129.94	Joback Method
cpg	1603.25	J/mol×K	1168.11	Joback Method
cpg	1641.50	J/mol×K	1206.29	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C17137701&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C17137701&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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