

# 1-Naphthalenecarboxylic acid, ethyl ester

<b>Other names:</b>	1-Naphthoic acid, ethyl ester Ethyl 1-naphthoate Ethyl «alpha»-naphthoate
<b>Inchi:</b>	InChI=1S/C13H12O2/c1-2-15-13(14)12-9-5-7-10-6-3-4-8-11(10)12/h3-9H,2H2,1H3
<b>InchiKey:</b>	XCTLDQQOHIEUCJ-UHFFFAOYSA-N
<b>Formula:</b>	C13H12O2
<b>SMILES:</b>	CCOC(=O)c1cccc2ccccc12
<b>Mol. weight [g/mol]:</b>	200.23
<b>CAS:</b>	3007-97-4

## Physical Properties

Property code	Value	Unit	Source
gf	34.09	kJ/mol	Joback Method
hf	-140.32	kJ/mol	Joback Method
hfus	22.88	kJ/mol	Joback Method
hvap	58.27	kJ/mol	Joback Method
log10ws	-3.94		Crippen Method
logp	3.017		Crippen Method
mcvol	158.250	ml/mol	McGowan Method
pc	2893.62	kPa	Joback Method
tb	623.77	K	Joback Method
tc	854.32	K	Joback Method
tf	380.07	K	Joback Method
vc	0.602	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	384.57	J/molxK	623.77	Joback Method
cpg	398.55	J/molxK	662.20	Joback Method
cpg	411.55	J/molxK	700.62	Joback Method
cpg	423.62	J/molxK	739.05	Joback Method
cpg	434.81	J/molxK	777.47	Joback Method
cpg	445.18	J/molxK	815.90	Joback Method

cpg	454.78	J/molxK	854.32	Joback Method
dvisc	0.0014189	Paxs	380.07	Joback Method
dvisc	0.0009480	Paxs	420.69	Joback Method
dvisc	0.0006800	Paxs	461.30	Joback Method
dvisc	0.0005147	Paxs	501.92	Joback Method
dvisc	0.0004062	Paxs	542.54	Joback Method
dvisc	0.0003313	Paxs	583.15	Joback Method
dvisc	0.0002775	Paxs	623.77	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3007974&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3007974&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/83-097-3/1-Naphthalenecarboxylic-acid-ethyl-ester.pdf>

Generated by Cheméo on 2024-04-19 01:46:20.56449121 +0000 UTC m=+15780429.485068521.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.