

2,6-Difluorobenzoic acid, hex-4-yn-3-yl ester

Inchi:	InChI=1S/C13H12F2O2/c1-3-6-9(4-2)17-13(16)12-10(14)7-5-8-11(12)15/h5,7-9H,4H2,1-
InchiKey:	MRLUKPZMKZZDKY-UHFFFAOYSA-N
Formula:	C13H12F2O2
SMILES:	CC#CC(CC)OC(=O)c1c(F)cccc1F
Mol. weight [g/mol]:	238.23

Physical Properties

Property code	Value	Unit	Source
gf	-271.45	kJ/mol	Joback Method
hf	-468.06	kJ/mol	Joback Method
hfus	31.24	kJ/mol	Joback Method
hvap	57.42	kJ/mol	Joback Method
log10ws	-4.38		Crippen Method
logp	2.923		Crippen Method
mcvol	172.650	ml/mol	McGowan Method
pc	2400.57	kPa	Joback Method
rinpola	1549.50		NIST Webbook
rinpola	1549.50		NIST Webbook
tb	616.87	K	Joback Method
tc	829.53	K	Joback Method
tf	452.17	K	Joback Method
vc	0.671	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	418.72	J/molxK	616.87	Joback Method
cpg	432.30	J/molxK	652.31	Joback Method
cpg	445.10	J/molxK	687.76	Joback Method
cpg	457.14	J/molxK	723.20	Joback Method
cpg	468.44	J/molxK	758.64	Joback Method
cpg	479.01	J/molxK	794.09	Joback Method
cpg	488.85	J/molxK	829.53	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U292584&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/83-101-7/2-6-Difluorobenzoic-acid-hex-4-yn-3-yl-ester.pdf>

Generated by Cheméo on 2024-04-26 07:06:43.882213169 +0000 UTC m=+16404452.802790482.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.