

# 2-((Pent-4-enyloxy)carbonyl)benzoic acid

<b>Inchi:</b>	InChI=1S/C13H14O4/c1-2-3-6-9-17-13(16)11-8-5-4-7-10(11)12(14)15/h2,4-5,7-8H,1,3,6,
<b>InchiKey:</b>	HYCQCFCIEMXEMJ-UHFFFAOYSA-N
<b>Formula:</b>	C13H14O4
<b>SMILES:</b>	C=CCCCOC(=O)c1ccccc1C(=O)O
<b>Mol. weight [g/mol]:</b>	234.25
<b>CAS:</b>	190184-82-8

## Physical Properties

Property code	Value	Unit	Source
gf	-250.46	kJ/mol	Joback Method
hf	-470.77	kJ/mol	Joback Method
hfus	30.27	kJ/mol	Joback Method
hvap	79.38	kJ/mol	Joback Method
log10ws	-3.30		Crippen Method
logp	2.508		Crippen Method
mcvol	180.850	ml/mol	McGowan Method
pc	2749.78	kPa	Joback Method
rinpol	1921.00		NIST Webbook
rinpol	1921.00		NIST Webbook
tb	747.52	K	Joback Method
tc	950.11	K	Joback Method
tf	456.36	K	Joback Method
vc	0.685	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	490.32	J/molxK	747.52	Joback Method
cpg	537.76	J/molxK	916.34	Joback Method
cpg	529.65	J/molxK	882.58	Joback Method
cpg	520.87	J/molxK	848.81	Joback Method
cpg	511.41	J/molxK	815.05	Joback Method
cpg	501.23	J/molxK	781.28	Joback Method
cpg	545.23	J/molxK	950.11	Joback Method

dvisc	0.0000377	Paxs	747.52	Joback Method
dvisc	0.0000536	Paxs	698.99	Joback Method
dvisc	0.0000802	Paxs	650.47	Joback Method
dvisc	0.0001282	Paxs	601.94	Joback Method
dvisc	0.0002226	Paxs	553.41	Joback Method
dvisc	0.0004296	Paxs	504.89	Joback Method
dvisc	0.0009534	Paxs	456.36	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C190184828&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C190184828&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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