

# N-(2,2-dimethylpropyl)-N-methyl-benzamide

<b>Inchi:</b>	InChI=1S/C13H19NO/c1-13(2,3)10-14(4)12(15)11-8-6-5-7-9-11/h5-9H,10H2,1-4H3
<b>InchiKey:</b>	NXLPXQWESLPQSU-UHFFFAOYSA-N
<b>Formula:</b>	C13H19NO
<b>SMILES:</b>	CN(CC(C)(C)C)C(=O)c1ccccc1
<b>Mol. weight [g/mol]:</b>	205.30

## Physical Properties

Property code	Value	Unit	Source
gf	155.69	kJ/mol	Joback Method
hf	-128.92	kJ/mol	Joback Method
hfus	20.67	kJ/mol	Joback Method
hvap	54.30	kJ/mol	Joback Method
log10ws	-3.05		Crippen Method
logp	2.805		Crippen Method
mcvol	181.820	ml/mol	McGowan Method
pc	2367.97	kPa	Joback Method
ripol	1637.79		NIST Webbook
ripol	1606.54		NIST Webbook
ripol	1657.39		NIST Webbook
ripol	1625.88		NIST Webbook
ripol	1606.54		NIST Webbook
ripol	2472.23		NIST Webbook
ripol	2510.89		NIST Webbook
ripol	2491.77		NIST Webbook
ripol	2472.23		NIST Webbook
tb	586.60	K	Joback Method
tc	800.78	K	Joback Method
tf	347.51	K	Joback Method
vc	0.668	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	456.88	J/molxK	586.60	Joback Method

cpg	474.44	J/mol×K	622.30	Joback Method
cpg	490.78	J/mol×K	657.99	Joback Method
cpg	505.97	J/mol×K	693.69	Joback Method
cpg	520.10	J/mol×K	729.39	Joback Method
cpg	533.23	J/mol×K	765.09	Joback Method
cpg	545.43	J/mol×K	800.78	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R194037&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R194037&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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