

# Undecanal dimethyl acetal

<b>Other names:</b>	Undecane, 1,1-dimethoxy-
<b>Inchi:</b>	InChI=1S/C13H28O2/c1-4-5-6-7-8-9-10-11-12-13(14-2)15-3/h13H,4-12H2,1-3H3
<b>InchiKey:</b>	FBJUQTUWWCVIDH-UHFFFAOYSA-N
<b>Formula:</b>	C13H28O2
<b>SMILES:</b>	CCCCCCCCCCC(OC)OC
<b>Mol. weight [g/mol]:</b>	216.36
<b>CAS:</b>	52517-67-6

## Physical Properties

Property code	Value	Unit	Source
gf	-153.86	kJ/mol	Joback Method
hf	-581.37	kJ/mol	Joback Method
hfus	28.28	kJ/mol	Joback Method
hvap	48.96	kJ/mol	Joback Method
log10ws	-4.05		Crippen Method
logp	4.136		Crippen Method
mcvol	205.770	ml/mol	McGowan Method
pc	1610.29	kPa	Joback Method
rinpola	1466.00		NIST Webbook
rinpola	1466.00		NIST Webbook
ripola	1668.00		NIST Webbook
ripola	1668.00		NIST Webbook
tb	541.24	K	Joback Method
tc	704.41	K	Joback Method
tf	265.73	K	Joback Method
vc	0.793	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	514.89	J/molxK	541.24	Joback Method
cpg	532.12	J/molxK	568.43	Joback Method
cpg	548.75	J/molxK	595.63	Joback Method
cpg	564.78	J/molxK	622.82	Joback Method

cpg	580.21	J/molxK	650.02	Joback Method
cpg	595.06	J/molxK	677.21	Joback Method
cpg	609.31	J/molxK	704.41	Joback Method
dvisc	0.0039976	Paxs	265.73	Joback Method
dvisc	0.0014601	Paxs	311.65	Joback Method
dvisc	0.0006907	Paxs	357.57	Joback Method
dvisc	0.0003874	Paxs	403.49	Joback Method
dvisc	0.0002446	Paxs	449.40	Joback Method
dvisc	0.0001681	Paxs	495.32	Joback Method
dvisc	0.0001232	Paxs	541.24	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C52517676&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C52517676&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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