

# 3-Hydroxy-«beta»-damascone

<b>Inchi:</b>	InChI=1S/C13H20O2/c1-5-6-11(15)12-9(2)10(14)7-8-13(12,3)4/h5-6,10,14H,7-8H2,1-4H
<b>InchiKey:</b>	NOORESDHJXLGAO-AATRIKPKSA-N
<b>Formula:</b>	C13H20O2
<b>SMILES:</b>	CC=CC(=O)C1=C(C)C(O)CCC1(C)C
<b>Mol. weight [g/mol]:</b>	208.30
<b>CAS:</b>	102488-09-5

## Physical Properties

Property code	Value	Unit	Source
gf	-104.99	kJ/mol	Joback Method
hf	-375.18	kJ/mol	Joback Method
hfus	22.37	kJ/mol	Joback Method
hvap	68.50	kJ/mol	Joback Method
log10ws	-3.28		Crippen Method
logp	2.629		Crippen Method
mcvol	182.010	ml/mol	McGowan Method
pc	2412.37	kPa	Joback Method
rinpol	1618.00		NIST Webbook
rinpol	1640.00		NIST Webbook
rinpol	1663.00		NIST Webbook
rinpol	1617.00		NIST Webbook
rinpol	1618.00		NIST Webbook
rinpol	1617.00		NIST Webbook
rinpol	1627.00		NIST Webbook
rinpol	1591.00		NIST Webbook
rinpol	1627.00		NIST Webbook
ripol	2535.00		NIST Webbook
ripol	2532.00		NIST Webbook
ripol	2559.00		NIST Webbook
ripol	2537.00		NIST Webbook
ripol	2537.00		NIST Webbook
ripol	2535.00		NIST Webbook
ripol	2546.00		NIST Webbook
ripol	2559.00		NIST Webbook
ripol	2544.00		NIST Webbook
ripol	2563.00		NIST Webbook
ripol	2537.00		NIST Webbook

ripol	2537.00		NIST Webbook
ripol	2525.00		NIST Webbook
ripol	2531.00		NIST Webbook
ripol	2563.00		NIST Webbook
ripol	2504.00		NIST Webbook
ripol	2504.00		NIST Webbook
ripol	2537.00		NIST Webbook
ripol	2532.00		NIST Webbook
ripol	2535.00		NIST Webbook
tb	671.29	K	Joback Method
tc	875.23	K	Joback Method
tf	394.78	K	Joback Method
vc	0.684	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	500.33	J/mol×K	671.29	Joback Method
cpg	515.41	J/mol×K	705.28	Joback Method
cpg	529.82	J/mol×K	739.27	Joback Method
cpg	543.66	J/mol×K	773.26	Joback Method
cpg	557.02	J/mol×K	807.25	Joback Method
cpg	570.02	J/mol×K	841.24	Joback Method
cpg	582.75	J/mol×K	875.23	Joback Method

## Sources

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C102488095&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

## Legend

**cpg:** Ideal gas heat capacity

<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>ripolar:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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