

# ethyl linalyl acetate 1

<b>Inchi:</b>	InChI=1S/C13H24O2/c1-6-13(7-2,15-12(5)14)10-8-9-11(3)4/h9H,6-8,10H2,1-5H3
<b>InchiKey:</b>	KVHMWAPYAFPSRE-UHFFFAOYSA-N
<b>Formula:</b>	C13H24O2
<b>SMILES:</b>	CCC(CC)(CCC=C(C)C)OC(C)=O
<b>Mol. weight [g/mol]:</b>	212.33

## Physical Properties

Property code	Value	Unit	Source
gf	-100.83	kJ/mol	Joback Method
hf	-457.77	kJ/mol	Joback Method
hfus	23.69	kJ/mol	Joback Method
hvap	52.43	kJ/mol	Joback Method
log10ws	-4.09		Crippen Method
logp	3.855		Crippen Method
mcvol	197.170	ml/mol	McGowan Method
pc	1820.06	kPa	Joback Method
ripol	1333.90		NIST Webbook
ripol	1316.50		NIST Webbook
ripol	1316.50		NIST Webbook
ripol	1653.00		NIST Webbook
ripol	1629.00		NIST Webbook
ripol	1629.00		NIST Webbook
tb	573.94	K	Joback Method
tc	761.04	K	Joback Method
tf	291.81	K	Joback Method
vc	0.757	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	498.53	J/molxK	573.94	Joback Method
cpg	515.55	J/molxK	605.12	Joback Method
cpg	531.70	J/molxK	636.31	Joback Method
cpg	547.02	J/molxK	667.49	Joback Method

cpg	561.54	J/mol×K	698.67	Joback Method
cpg	575.30	J/mol×K	729.85	Joback Method
cpg	588.34	J/mol×K	761.04	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R185503&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R185503&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/83-279-1/ethyl-linalyl-acetate-1.pdf>

Generated by Cheméo on 2024-04-26 14:06:21.905399433 +0000 UTC m=+16429630.825976748.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.