

Butane, 2,2-dichloro-3-methyl-

Other names:	2,2-Dichloro-3-methylbutane
Inchi:	InChI=1S/C5H10Cl2/c1-4(2)5(3,6)7/h4H,1-3H3
InchiKey:	WIQMOFSSJHPXIK-UHFFFAOYSA-N
Formula:	C5H10Cl2
SMILES:	CC(C)C(C)(Cl)Cl
Mol. weight [g/mol]:	141.04
CAS:	17773-66-9

Physical Properties

Property code	Value	Unit	Source
gf	-32.24	kJ/mol	Joback Method
hf	-192.04	kJ/mol	Joback Method
hfus	6.16	kJ/mol	Joback Method
hvap	33.81	kJ/mol	Joback Method
log10ws	-2.59		Crippen Method
logp	2.836		Crippen Method
mcvol	105.790	ml/mol	McGowan Method
pc	3284.05	kPa	Joback Method
tb	384.99	K	Joback Method
tc	583.40	K	Joback Method
tf	193.37	K	Joback Method
vc	0.397	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	177.35	J/molxK	384.99	Joback Method
cpg	187.87	J/molxK	418.06	Joback Method
cpg	197.76	J/molxK	451.13	Joback Method
cpg	207.03	J/molxK	484.19	Joback Method
cpg	215.73	J/molxK	517.26	Joback Method
cpg	223.89	J/molxK	550.33	Joback Method
cpg	231.52	J/molxK	583.40	Joback Method
dvisc	0.0136726	Paxs	193.37	Joback Method

dvisc	0.0048915	Paxs	225.31	Joback Method
dvisc	0.0022588	Paxs	257.24	Joback Method
dvisc	0.0012372	Paxs	289.18	Joback Method
dvisc	0.0007638	Paxs	321.12	Joback Method
dvisc	0.0005146	Paxs	353.05	Joback Method
dvisc	0.0003701	Paxs	384.99	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.76655e+01
Coeff. B	-4.68146e+03
Coeff. C	-6.39290e+01
Temperature range (K), min.	333.32
Temperature range (K), max.	442.87

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C17773669&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure

Legend

cp_g:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h_{vap}:	Enthalpy of vaporization at standard conditions

log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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