

2,5-Dimethoxyphenethylamine

Inchi:	InChI=1S/C10H15NO2/c1-12-9-3-4-10(13-2)8(7-9)5-6-11/h3-4,7H,5-6,11H2,1-2H3
InchiKey:	WNCUVUUEJZEATP-UHFFFAOYSA-N
Formula:	C10H15NO2
SMILES:	COc1ccc(OC)c(CCN)c1
Mol. weight [g/mol]:	181.23
CAS:	3600-86-0

Physical Properties

Property code	Value	Unit	Source
gf	-17.08	kJ/mol	Joback Method
hf	-266.79	kJ/mol	Joback Method
hfus	22.49	kJ/mol	Joback Method
hvap	56.92	kJ/mol	Joback Method
log10ws	-1.95		Crippen Method
logp	1.205		Crippen Method
mcvol	149.720	ml/mol	McGowan Method
pc	2909.25	kPa	Joback Method
tb	582.21	K	Joback Method
tc	795.30	K	Joback Method
tf	381.64	K	Joback Method
vc	0.552	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	364.65	J/molxK	582.21	Joback Method
cpg	378.45	J/molxK	617.72	Joback Method
cpg	391.58	J/molxK	653.24	Joback Method
cpg	404.04	J/molxK	688.75	Joback Method
cpg	415.81	J/molxK	724.27	Joback Method
cpg	426.90	J/molxK	759.78	Joback Method
cpg	437.30	J/molxK	795.30	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C3600860&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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