

# 4,4-Dimethyl-2-imidazoline

<b>Inchi:</b>	InChI=1S/C5H10N2/c1-5(2)3-6-4-7-5/h4H,3H2,1-2H3,(H,6,7)
<b>InchiKey:</b>	OVABIFHEPZPSCK-UHFFFAOYSA-N
<b>Formula:</b>	C5H10N2
<b>SMILES:</b>	CC1(C)CN=CN1
<b>Mol. weight [g/mol]:</b>	98.15
<b>CAS:</b>	2305-59-1

## Physical Properties

Property code	Value	Unit	Source
affp	988.10	kJ/mol	NIST Webbook
basg	955.70	kJ/mol	NIST Webbook
gf	256.73	kJ/mol	Joback Method
hf	95.75	kJ/mol	Joback Method
hfus	12.29	kJ/mol	Joback Method
hvap	39.09	kJ/mol	Joback Method
log10ws	-0.77		Crippen Method
logp	0.396		Crippen Method
mcvol	86.110	ml/mol	McGowan Method
pc	5022.80	kPa	Joback Method
tb	430.73	K	Joback Method
tc	664.32	K	Joback Method
tf	358.24	K	Joback Method
vc	0.327	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	175.45	J/mol×K	430.73	Joback Method
cpg	189.57	J/mol×K	469.66	Joback Method
cpg	202.66	J/mol×K	508.59	Joback Method
cpg	214.81	J/mol×K	547.52	Joback Method
cpg	226.13	J/mol×K	586.46	Joback Method
cpg	236.72	J/mol×K	625.39	Joback Method
cpg	246.70	J/mol×K	664.32	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	355.70	K	2.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2305591&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2305591&amp;Units=SI</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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