

3-(Methylthio)-2-butanone

Other names:	2-Butanone, 3-(methylthio)- dl-3-(Methylthio)-2-butanone DL-3-(methylthio)butanone
Inchi:	InChI=1S/C5H10OS/c1-4(6)5(2)7-3/h5H,1-3H3
InchiKey:	HFVLNCDRAMUMCC-UHFFFAOYSA-N
Formula:	C5H10OS
SMILES:	CSC(C)C(C)=O
Mol. weight [g/mol]:	118.20
CAS:	53475-15-3

Physical Properties

Property code	Value	Unit	Source
gf	-107.02	kJ/mol	Joback Method
hf	-222.52	kJ/mol	Joback Method
hfus	10.91	kJ/mol	Joback Method
hvap	39.90	kJ/mol	Joback Method
log10ws	-1.19		Crippen Method
logp	1.327		Crippen Method
mcvol	99.230	ml/mol	McGowan Method
pc	3857.88	kPa	Joback Method
tb	436.01	K	Joback Method
tc	644.03	K	Joback Method
tf	215.44	K	Joback Method
vc	0.369	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	184.66	J/mol×K	436.01	Joback Method
cpg	194.48	J/mol×K	470.68	Joback Method
cpg	203.89	J/mol×K	505.35	Joback Method
cpg	212.87	J/mol×K	540.02	Joback Method
cpg	221.43	J/mol×K	574.69	Joback Method
cpg	229.58	J/mol×K	609.36	Joback Method

cpg

237.31

J/mol×K

644.03

Joback Method

Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	325.20	K	2.70	NIST Webbook

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C53475153&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tbrp:	Boiling point at reduced pressure
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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