

iso-Propyl N-methyl anthranilate

Inchi:	InChI=1S/C11H15NO2/c1-8(2)14-11(13)9-6-4-5-7-10(9)12-3/h4-8,12H,1-3H3
InchiKey:	OBKVPPPSOAFJDS-UHFFFAOYSA-N
Formula:	C11H15NO2
SMILES:	CNc1ccccc1C(=O)OC(C)C
Mol. weight [g/mol]:	193.24
CAS:	99985-64-5

Physical Properties

Property code	Value	Unit	Source
gf	-2.45	kJ/mol	Joback Method
hf	-241.92	kJ/mol	Joback Method
hfus	22.26	kJ/mol	Joback Method
hvap	58.22	kJ/mol	Joback Method
log10ws	-2.67		Crippen Method
logp	2.293		Crippen Method
mcvol	159.510	ml/mol	McGowan Method
pc	2796.51	kPa	Joback Method
rinpol	1491.00		NIST Webbook
rinpol	1491.00		NIST Webbook
ripol	2058.00		NIST Webbook
ripol	2058.00		NIST Webbook
tb	608.76	K	Joback Method
tc	823.24	K	Joback Method
tf	362.49	K	Joback Method
vc	0.597	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	396.17	J/molxK	608.76	Joback Method
cpg	410.54	J/molxK	644.51	Joback Method
cpg	424.06	J/molxK	680.25	Joback Method
cpg	436.76	J/molxK	716.00	Joback Method
cpg	448.64	J/molxK	751.75	Joback Method

cpg	459.72	J/mol×K	787.50	Joback Method
cpg	470.03	J/mol×K	823.24	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C99985645&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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