

# 2-(3,4-Dichlorophenoxy)ethyl 2,2-dichloropropanoate

Inchi:	InChI=1S/C11H10Cl4O3/c1-11(14,15)10(16)18-5-4-17-7-2-3-8(12)9(13)6-7/h2-3,6H,4-5H
InchiKey:	QCMMKKFIKJOVEU-UHFFFAOYSA-N
Formula:	C11H10Cl4O3
SMILES:	CC(Cl)(Cl)C(=O)OCCOc1ccc(Cl)c(Cl)c1
Mol. weight [g/mol]:	332.01
CAS:	116402-59-6

## Physical Properties

Property code	Value	Unit	Source
gf	-248.91	kJ/mol	Joback Method
hf	-505.51	kJ/mol	Joback Method
hfus	30.86	kJ/mol	Joback Method
hvap	71.49	kJ/mol	Joback Method
log10ws	-4.41		Crippen Method
logp	4.109		Crippen Method
mvol	204.360	ml/mol	McGowan Method
pc	2327.03	kPa	Joback Method
tb	732.92	K	Joback Method
tc	967.10	K	Joback Method
tf	481.68	K	Joback Method
vc	0.770	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	468.78	J/molxK	732.92	Joback Method
cpg	511.88	J/molxK	928.07	Joback Method
cpg	504.90	J/molxK	889.04	Joback Method
cpg	497.12	J/molxK	850.01	Joback Method
cpg	488.53	J/molxK	810.98	Joback Method
cpg	479.09	J/molxK	771.95	Joback Method
cpg	518.11	J/molxK	967.10	Joback Method
dvisc	0.0000933	Paxs	732.92	Joback Method
dvisc	0.0001165	Paxs	691.05	Joback Method

dvisc	0.0001498	Paxs	649.17	Joback Method
dvisc	0.0001994	Paxs	607.30	Joback Method
dvisc	0.0002770	Paxs	565.43	Joback Method
dvisc	0.0004054	Paxs	523.55	Joback Method
dvisc	0.0006340	Paxs	481.68	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C116402596&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C116402596&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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