

# 5-Chlorovaleric acid, 2-adamantyl ester

<b>Inchi:</b>	InChI=1S/C15H23ClO2/c16-4-2-1-3-14(17)18-15-12-6-10-5-11(8-12)9-13(15)7-10/h10-13
<b>InchiKey:</b>	XEDXKFLTKHZQND-UHFFFAOYSA-N
<b>Formula:</b>	C15H23ClO2
<b>SMILES:</b>	O=C(CCCCCl)OC1C2CC3CC(C2)CC1C3
<b>Mol. weight [g/mol]:</b>	270.80

## Physical Properties

Property code	Value	Unit	Source
gf	-15.70	kJ/mol	Joback Method
hf	-441.91	kJ/mol	Joback Method
hfus	36.04	kJ/mol	Joback Method
hvap	61.82	kJ/mol	Joback Method
log10ws	-3.95		Crippen Method
logp	3.763		Crippen Method
mvol	209.310	ml/mol	McGowan Method
pc	1890.36	kPa	Joback Method
rinpol	2055.20		NIST Webbook
rinpol	2055.20		NIST Webbook
tb	671.47	K	Joback Method
tc	879.31	K	Joback Method
tf	402.71	K	Joback Method
vc	0.809	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	618.94	J/molxK	671.47	Joback Method
cpg	705.02	J/molxK	844.67	Joback Method
cpg	689.88	J/molxK	810.03	Joback Method
cpg	673.78	J/molxK	775.39	Joback Method
cpg	656.65	J/molxK	740.75	Joback Method
cpg	638.39	J/molxK	706.11	Joback Method
cpg	719.29	J/molxK	879.31	Joback Method
dvisc	0.0023890	Paxs	671.47	Joback Method

dvisc	0.0025016	Paxs	626.68	Joback Method
dvisc	0.0026383	Paxs	581.88	Joback Method
dvisc	0.0028071	Paxs	537.09	Joback Method
dvisc	0.0030207	Paxs	492.30	Joback Method
dvisc	0.0032987	Paxs	447.50	Joback Method
dvisc	0.0036734	Paxs	402.71	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U292243&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U292243&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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