

# Myraldyl acetate 1

<b>Inchi:</b>	InChI=1S/C15H24O2/c1-12(2)6-4-7-14-8-5-9-15(10-14)11-17-13(3)16/h6,8,15H,4-5,7,9-11
<b>InchiKey:</b>	IIUKCYITROTKFB-UHFFFAOYSA-N
<b>Formula:</b>	C15H24O2
<b>SMILES:</b>	CC(=O)OCC1CCC=C(CCC=C(C)C)C1
<b>Mol. weight [g/mol]:</b>	236.35

## Physical Properties

Property code	Value	Unit	Source
gf	-42.05	kJ/mol	Joback Method
hf	-389.67	kJ/mol	Joback Method
hfus	28.95	kJ/mol	Joback Method
hvap	59.56	kJ/mol	Joback Method
log10ws	-4.32		Crippen Method
logp	4.022		Crippen Method
mcvol	210.190	ml/mol	McGowan Method
pc	1843.58	kPa	Joback Method
rinpol	1704.00		NIST Webbook
rinpol	1682.90		NIST Webbook
ripol	2166.00		NIST Webbook
ripol	2199.00		NIST Webbook
tb	646.62	K	Joback Method
tc	851.38	K	Joback Method
tf	332.59	K	Joback Method
vc	0.799	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	571.56	J/molxK	646.62	Joback Method
cpg	590.53	J/molxK	680.75	Joback Method
cpg	608.43	J/molxK	714.87	Joback Method
cpg	625.29	J/molxK	749.00	Joback Method
cpg	641.15	J/molxK	783.13	Joback Method
cpg	656.03	J/molxK	817.26	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R185865&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R185865&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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