

# Succinic acid, 2-methylpent-3-yl cyclopentyl ester

Inchi:	InChI=1S/C15H26O4/c1-4-13(11(2)3)19-15(17)10-9-14(16)18-12-7-5-6-8-12/h11-13H,4-
InchiKey:	HYH DUJKNZZERDN-UHFFFAOYSA-N
Formula:	C15H26O4
SMILES:	CCC(OC(=O)CCC(=O)OC1CCCC1)C(C)C
Mol. weight [g/mol]:	270.36

## Physical Properties

Property code	Value	Unit	Source
gf	-360.75	kJ/mol	Joback Method
hf	-792.61	kJ/mol	Joback Method
hfus	27.07	kJ/mol	Joback Method
hvap	66.78	kJ/mol	Joback Method
log10ws	-3.70		Crippen Method
logp	3.230		Crippen Method
mcvol	226.230	ml/mol	McGowan Method
pc	1780.34	kPa	Joback Method
rinsol	1807.00		NIST Webbook
tb	709.58	K	Joback Method
tc	908.41	K	Joback Method
tf	384.03	K	Joback Method
vc	0.853	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	669.69	J/molxK	709.58	Joback Method
cpg	687.76	J/molxK	742.72	Joback Method
cpg	704.75	J/molxK	775.86	Joback Method
cpg	720.68	J/molxK	809.00	Joback Method
cpg	735.57	J/molxK	842.13	Joback Method
cpg	749.42	J/molxK	875.27	Joback Method
cpg	762.25	J/molxK	908.41	Joback Method
dvisc	0.0024150	Paxs	384.03	Joback Method
dvisc	0.0010977	Paxs	438.29	Joback Method

dvisc	0.0005936	Paxs	492.55	Joback Method
dvisc	0.0003627	Paxs	546.81	Joback Method
dvisc	0.0002422	Paxs	601.06	Joback Method
dvisc	0.0001729	Paxs	655.32	Joback Method
dvisc	0.0001300	Paxs	709.58	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U391374&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U391374&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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