

# Glutaric acid, oct-1-en-3-yl 2,2-dichloroethyl ester

Inchi:	InChI=1S/C15H24Cl2O4/c1-3-5-6-8-12(4-2)21-15(19)10-7-9-14(18)20-11-13(16)17/h4,12
InchiKey:	OWVDYXTZXPBCRM-UHFFFAOYSA-N
Formula:	C15H24Cl2O4
SMILES:	C=CC(CCCCC)OC(=O)CCCC(=O)OCC(Cl)Cl
Mol. weight [g/mol]:	339.25

## Physical Properties

Property code	Value	Unit	Source
gf	-333.32	kJ/mol	Joback Method
hf	-759.14	kJ/mol	Joback Method
hfus	40.25	kJ/mol	Joback Method
hvap	74.62	kJ/mol	Joback Method
log10ws	-4.71		Crippen Method
logp	4.182		Crippen Method
mcvol	257.270	ml/mol	McGowan Method
pc	1499.99	kPa	Joback Method
rinpola	2063.00		NIST Webbook
rinpola	2063.00		NIST Webbook
tb	765.84	K	Joback Method
tc	957.29	K	Joback Method
tf	431.21	K	Joback Method
vc	0.991	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	712.73	J/molxK	765.84	Joback Method
cpg	726.93	J/molxK	797.75	Joback Method
cpg	740.26	J/molxK	829.66	Joback Method
cpg	752.73	J/molxK	861.56	Joback Method
cpg	764.37	J/molxK	893.47	Joback Method
cpg	775.18	J/molxK	925.38	Joback Method
cpg	785.19	J/molxK	957.29	Joback Method
dvisc	0.0012673	Paxs	431.21	Joback Method

dvisc	0.0006005	Paxs	486.98	Joback Method
dvisc	0.0003317	Paxs	542.75	Joback Method
dvisc	0.0002047	Paxs	598.52	Joback Method
dvisc	0.0001371	Paxs	654.30	Joback Method
dvisc	0.0000978	Paxs	710.07	Joback Method
dvisc	0.0000733	Paxs	765.84	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405348&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405348&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/83-702-0/Glutaric-acid-oct-1-en-3-yl-2-2-dichloroethyl-ester.pdf>

Generated by Cheméo on 2024-04-30 07:08:09.14606261 +0000 UTC m=+16750138.066639922.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.