

# Carbonic acid, monoamide, N-2-ethylhexyl-, hexyl ester

Inchi:	InChI=1S/C15H31NO2/c1-4-7-9-10-12-18-15(17)16-13-14(6-3)11-8-5-2/h14H,4-13H2,1-3
InchiKey:	VQSCCKZPLAJIHAS-UHFFFAOYSA-N
Formula:	C15H31NO2
SMILES:	CCCCCCCOC(O)=NCC(CC)CCCC
Mol. weight [g/mol]:	257.41

## Physical Properties

Property code	Value	Unit	Source
hf	-570.23	kJ/mol	Joback Method
hvap	71.08	kJ/mol	Joback Method
log10ws	-4.43		Crippen Method
logp	4.714		Crippen Method
mcvol	239.630	ml/mol	McGowan Method
pc	1397.50	kPa	Joback Method
rinpol	1834.00		NIST Webbook
tb	733.32	K	Joback Method
tc	910.48	K	Joback Method

## Sources

NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406711&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406711&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l

<b>logP:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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