

# Isophthalic acid, 4-isopropylphenyl octyl ester

<b>Inchi:</b>	InChI=1S/C25H32O4/c1-4-5-6-7-8-9-17-28-24(26)21-11-10-12-22(18-21)25(27)29-23-15
<b>InchiKey:</b>	SLOXNMDPFUBJEL-UHFFFAOYSA-N
<b>Formula:</b>	C25H32O4
<b>SMILES:</b>	CCCCCCCCOC(=O)c1cccc(C(=O)Oc2ccc(C(C)C)cc2)c1
<b>Mol. weight [g/mol]:</b>	396.52

## Physical Properties

Property code	Value	Unit	Source
gf	-105.10	kJ/mol	Joback Method
hf	-604.09	kJ/mol	Joback Method
hfus	49.86	kJ/mol	Joback Method
hvap	95.04	kJ/mol	Joback Method
log10ws	-7.92		Crippen Method
logp	6.546		Crippen Method
mvol	330.470	ml/mol	McGowan Method
pc	1200.62	kPa	Joback Method
rinpol	3206.00		NIST Webbook
rinpol	3206.00		NIST Webbook
tb	986.86	K	Joback Method
tc	1213.38	K	Joback Method
tf	578.71	K	Joback Method
vc	1.262	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1076.37	J/molxK	986.86	Joback Method
cpg	1090.75	J/molxK	1024.61	Joback Method
cpg	1103.65	J/molxK	1062.37	Joback Method
cpg	1115.13	J/molxK	1100.12	Joback Method
cpg	1125.22	J/molxK	1137.87	Joback Method
cpg	1134.00	J/molxK	1175.63	Joback Method
cpg	1141.50	J/molxK	1213.38	Joback Method
dvisc	0.0002911	Paxs	578.71	Joback Method

dvisc	0.0001568	Paxs	646.74	Joback Method
dvisc	0.0000951	Paxs	714.76	Joback Method
dvisc	0.0000628	Paxs	782.79	Joback Method
dvisc	0.0000444	Paxs	850.81	Joback Method
dvisc	0.0000330	Paxs	918.84	Joback Method
dvisc	0.0000256	Paxs	986.86	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U344459&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U344459&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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