

# 3-Cyclopentylpropionic acid, 3-chloroprop-2-enyl ester

Inchi:	InChI=1S/C11H17ClO2/c12-8-3-9-14-11(13)7-6-10-4-1-2-5-10/h3,8,10H,1-2,4-7,9H2/b8-3
InchiKey:	QXWNSXGSUGCIHO-FPYGCLRLSA-N
Formula:	C11H17ClO2
SMILES:	O=C(CCC1CCCC1)OCC=CCl
Mol. weight [g/mol]:	216.70

## Physical Properties

Property code	Value	Unit	Source
gf	-87.34	kJ/mol	Joback Method
hf	-353.21	kJ/mol	Joback Method
hfus	25.37	kJ/mol	Joback Method
hvap	53.84	kJ/mol	Joback Method
log10ws	-3.44		Crippen Method
logp	3.252		Crippen Method
mvol	170.370	ml/mol	McGowan Method
pc	2419.50	kPa	Joback Method
rinpol	1555.10		NIST Webbook
rinpol	1555.10		NIST Webbook
tb	584.24	K	Joback Method
tc	792.73	K	Joback Method
tf	321.63	K	Joback Method
vc	0.645	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	415.39	J/molxK	584.24	Joback Method
cpg	431.62	J/molxK	618.99	Joback Method
cpg	446.89	J/molxK	653.74	Joback Method
cpg	461.25	J/molxK	688.48	Joback Method
cpg	474.73	J/molxK	723.23	Joback Method
cpg	487.37	J/molxK	757.98	Joback Method
cpg	499.21	J/molxK	792.73	Joback Method
dvisc	0.0027862	Paxs	321.63	Joback Method

dvisc	0.0014396	Paxs	365.40	Joback Method
dvisc	0.0008567	Paxs	409.17	Joback Method
dvisc	0.0005636	Paxs	452.94	Joback Method
dvisc	0.0003992	Paxs	496.70	Joback Method
dvisc	0.0002990	Paxs	540.47	Joback Method
dvisc	0.0002338	Paxs	584.24	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U292472&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U292472&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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