

# cis-Cyclohex-4-en-1,2-dicarboxylic acid, hexyl undecyl ester

Inchi:	InChI=1S/C25H44O4/c1-3-5-7-9-10-11-12-13-17-21-29-25(27)23-19-15-14-18-22(23)24(
InchiKey:	VACVHLOQJNPQOU-UHFFFAOYSA-N
Formula:	C25H44O4
SMILES:	CCCCCCCCCOC(=O)C1CC=CCC1C(=O)OCCCCC
Mol. weight [g/mol]:	408.61

## Physical Properties

Property code	Value	Unit	Source
gf	-261.52	kJ/mol	Joback Method
hf	-957.17	kJ/mol	Joback Method
hfus	60.21	kJ/mol	Joback Method
hvap	89.97	kJ/mol	Joback Method
log10ws	-7.28		Crippen Method
logp	6.766		Crippen Method
mcvol	362.830	ml/mol	McGowan Method
pc	908.34	kPa	Joback Method
rinpol	2781.00		NIST Webbook
tb	938.02	K	Joback Method
tc	1148.40	K	Joback Method
tf	519.73	K	Joback Method
vc	1.401	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1251.69	J/molxK	938.02	Joback Method
cpg	1270.92	J/molxK	973.08	Joback Method
cpg	1288.52	J/molxK	1008.15	Joback Method
cpg	1304.50	J/molxK	1043.21	Joback Method
cpg	1318.92	J/molxK	1078.27	Joback Method
cpg	1331.81	J/molxK	1113.34	Joback Method
cpg	1343.20	J/molxK	1148.40	Joback Method
dvisc	0.0005756	Paxs	519.73	Joback Method
dvisc	0.0002790	Paxs	589.45	Joback Method

dvisc	0.0001577	Paxs	659.16	Joback Method
dvisc	0.0000994	Paxs	728.88	Joback Method
dvisc	0.0000679	Paxs	798.59	Joback Method
dvisc	0.0000493	Paxs	868.31	Joback Method
dvisc	0.0000375	Paxs	938.02	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U382668&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U382668&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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