

# Isophthalic acid, 2-biphenyl pentyl ester

<b>Inchi:</b>	InChI=1S/C25H24O4/c1-2-3-9-17-28-24(26)20-13-10-14-21(18-20)25(27)29-23-16-8-7-1
<b>InchiKey:</b>	RXKFPALSYXIH LG-UHFFFAOYSA-N
<b>Formula:</b>	C25H24O4
<b>SMILES:</b>	CCCCCOC(=O)c1cccc(C(=O)Oc2ccccc2-c2ccccc2)c1
<b>Mol. weight [g/mol]:</b>	388.46

## Physical Properties

Property code	Value	Unit	Source
gf	9.75	kJ/mol	Joback Method
hf	-362.28	kJ/mol	Joback Method
hfus	47.42	kJ/mol	Joback Method
hvap	97.71	kJ/mol	Joback Method
log10ws	-8.08		Crippen Method
logp	5.920		Crippen Method
mvol	306.710	ml/mol	McGowan Method
pc	1522.31	kPa	Joback Method
rinpol	3119.00		NIST Webbook
rinpol	3119.00		NIST Webbook
tb	1013.98	K	Joback Method
tc	1258.41	K	Joback Method
tf	620.13	K	Joback Method
vc	1.159	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	966.92	J/molxK	1013.98	Joback Method
cpg	1012.17	J/molxK	1217.68	Joback Method
cpg	1005.81	J/molxK	1176.94	Joback Method
cpg	998.17	J/molxK	1136.20	Joback Method
cpg	989.20	J/molxK	1095.46	Joback Method
cpg	978.81	J/molxK	1054.72	Joback Method
cpg	1017.34	J/molxK	1258.41	Joback Method
dvisc	0.0000292	Paxs	1013.98	Joback Method

dvisc	0.0000368	Paxs	948.34	Joback Method
dvisc	0.0000481	Paxs	882.70	Joback Method
dvisc	0.0000655	Paxs	817.06	Joback Method
dvisc	0.0000942	Paxs	751.41	Joback Method
dvisc	0.0001453	Paxs	685.77	Joback Method
dvisc	0.0002455	Paxs	620.13	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U344560&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U344560&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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