

# Fumaric acid, 3,5-dimethylphenyl dodec-2-en-1-yl ester

<b>Inchi:</b>	InChI=1S/C24H34O4/c1-4-5-6-7-8-9-10-11-12-13-16-27-23(25)14-15-24(26)28-22-18-20
<b>InchiKey:</b>	AEFAUOCCFJSWSU-SQIWNDDBBSA-N
<b>Formula:</b>	C24H34O4
<b>SMILES:</b>	CCCCCCCCC=CCOC(=O)C=CC(=O)Oc1cc(C)cc(C)c1
<b>Mol. weight [g/mol]:</b>	386.52

## Physical Properties

Property code	Value	Unit	Source
gf	-63.05	kJ/mol	Joback Method
hf	-580.26	kJ/mol	Joback Method
hfus	57.16	kJ/mol	Joback Method
hvap	90.85	kJ/mol	Joback Method
log10ws	-7.17		Crippen Method
logp	6.005		Crippen Method
mvol	331.540	ml/mol	McGowan Method
pc	1092.82	kPa	Joback Method
rinpol	2903.00		NIST Webbook
rinpol	2903.00		NIST Webbook
tb	946.06	K	Joback Method
tc	1160.68	K	Joback Method
tf	545.86	K	Joback Method
vc	1.280	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1062.77	J/molxK	946.06	Joback Method
cpg	1078.98	J/molxK	981.83	Joback Method
cpg	1094.05	J/molxK	1017.60	Joback Method
cpg	1108.04	J/molxK	1053.37	Joback Method
cpg	1121.00	J/molxK	1089.14	Joback Method
cpg	1132.99	J/molxK	1124.91	Joback Method
cpg	1144.08	J/molxK	1160.68	Joback Method
dvisc	0.0002863	Paxs	545.86	Joback Method

dvisc	0.0001515	Paxs	612.56	Joback Method
dvisc	0.0000909	Paxs	679.26	Joback Method
dvisc	0.0000597	Paxs	745.96	Joback Method
dvisc	0.0000420	Paxs	812.66	Joback Method
dvisc	0.0000312	Paxs	879.36	Joback Method
dvisc	0.0000242	Paxs	946.06	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405746&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405746&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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