

# Fumaric acid, pentadecyl pent-4-enyl ester

<b>Inchi:</b>	InChI=1S/C24H42O4/c1-3-5-7-8-9-10-11-12-13-14-15-16-18-22-28-24(26)20-19-23(25)2
<b>InchiKey:</b>	AEPHNMFYDIXXEI-FMQUCBEESA-N
<b>Formula:</b>	C24H42O4
<b>SMILES:</b>	C=CCCCOC(=O)C=CC(=O)OCCCCCCCCCCCCCCC
<b>Mol. weight [g/mol]:</b>	394.59

## Physical Properties

Property code	Value	Unit	Source
gf	-148.58	kJ/mol	Joback Method
hf	-785.64	kJ/mol	Joback Method
hfus	62.41	kJ/mol	Joback Method
hvap	86.62	kJ/mol	Joback Method
log10ws	-7.30		Crippen Method
logp	6.686		Crippen Method
mcvol	355.300	ml/mol	McGowan Method
pc	898.56	kPa	Joback Method
rinqol	2777.00		NIST Webbook
tb	901.94	K	Joback Method
tc	1104.44	K	Joback Method
tf	497.72	K	Joback Method
vc	1.389	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1161.65	J/molxK	901.94	Joback Method
cpg	1244.83	J/molxK	1070.69	Joback Method
cpg	1230.44	J/molxK	1036.94	Joback Method
cpg	1214.98	J/molxK	1003.19	Joback Method
cpg	1198.39	J/molxK	969.44	Joback Method
cpg	1180.63	J/molxK	935.69	Joback Method
cpg	1258.20	J/molxK	1104.44	Joback Method
dvisc	0.0000277	Paxs	901.94	Joback Method
dvisc	0.0000369	Paxs	834.57	Joback Method

dvisc	0.0000518	Paxs	767.20	Joback Method
dvisc	0.0000775	Paxs	699.83	Joback Method
dvisc	0.0001265	Paxs	632.46	Joback Method
dvisc	0.0002320	Paxs	565.09	Joback Method
dvisc	0.0005015	Paxs	497.72	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348855&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348855&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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