

# Fumaric acid, di(10-chlorodecyl) ester

**Inchi:** InChI=1S/C24H42Cl2O4/c25-19-13-9-5-1-3-7-11-15-21-29-23(27)17-18-24(28)30-22-16-  
**InchiKey:** QQCGGPBFZAAEAI-ISLYRVAYSA-N  
**Formula:** C24H42Cl2O4  
**SMILES:** O=C(C=CC(=O)OCCCCCCCCCCCCI)OCCCCCCCCCCCCI  
**Mol. weight [g/mol]:** 465.49

## Physical Properties

Property code	Value	Unit	Source
gf	-260.28	kJ/mol	Joback Method
hf	-942.55	kJ/mol	Joback Method
hfus	72.09	kJ/mol	Joback Method
hvap	96.06	kJ/mol	Joback Method
log10ws	-7.75		Crippen Method
logp	7.348		Crippen Method
mvol	384.080	ml/mol	McGowan Method
pc	839.19	kPa	Joback Method
rinpol	3469.00		NIST Webbook
rinpol	3469.00		NIST Webbook
tb	980.12	K	Joback Method
tc	1203.91	K	Joback Method
tf	559.32	K	Joback Method
vc	1.506	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1248.82	J/molxK	980.12	Joback Method
cpg	1266.67	J/molxK	1017.42	Joback Method
cpg	1283.14	J/molxK	1054.72	Joback Method
cpg	1298.28	J/molxK	1092.02	Joback Method
cpg	1312.16	J/molxK	1129.31	Joback Method
cpg	1324.85	J/molxK	1166.61	Joback Method
cpg	1336.42	J/molxK	1203.91	Joback Method
dvisc	0.0002915	Paxs	559.32	Joback Method

dvisc	0.0001416	Paxs	629.45	Joback Method
dvisc	0.0000795	Paxs	699.59	Joback Method
dvisc	0.0000496	Paxs	769.72	Joback Method
dvisc	0.0000334	Paxs	839.85	Joback Method
dvisc	0.0000240	Paxs	909.99	Joback Method
dvisc	0.0000180	Paxs	980.12	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348315&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348315&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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