

# Carbonic acid, (1R)-(-)-menthyl tridecyl ester

**Inchi:** InChI=1S/C24H46O3/c1-5-6-7-8-9-10-11-12-13-14-15-18-26-24(25)27-23-19-22(20(2)3)  
**InchiKey:** HENCOCBUHCDCOT-UHFFFAOYSA-N  
**Formula:** C24H46O3  
**SMILES:** CCCCCCCCCCCCCOC(=O)OC1CC(C(C)C)CCC1C  
**Mol. weight [g/mol]:** 382.62

## Physical Properties

Property code	Value	Unit	Source
gf	-181.13	kJ/mol	Joback Method
hf	-907.35	kJ/mol	Joback Method
hfus	52.34	kJ/mol	Joback Method
hvap	80.01	kJ/mol	Joback Method
log10ws	-8.08		Crippen Method
logp	7.911		Crippen Method
mcvol	351.470	ml/mol	McGowan Method
pc	893.20	kPa	Joback Method
rinpol	2606.00		NIST Webbook
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tb	857.00	K	Joback Method
tc	1051.78	K	Joback Method
tf	438.53	K	Joback Method
vc	1.347	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1196.43	J/molxK	857.00	Joback Method
cpg	1291.88	J/molxK	1019.32	Joback Method
cpg	1275.76	J/molxK	986.85	Joback Method
cpg	1258.17	J/molxK	954.39	Joback Method
cpg	1239.11	J/molxK	921.93	Joback Method
cpg	1218.53	J/molxK	889.46	Joback Method
cpg	1306.56	J/molxK	1051.78	Joback Method
dvisc	0.0000496	Paxs	857.00	Joback Method

dvisc	0.0000657	Paxs	787.25	Joback Method
dvisc	0.0000920	Paxs	717.51	Joback Method
dvisc	0.0001385	Paxs	647.76	Joback Method
dvisc	0.0002302	Paxs	578.02	Joback Method
dvisc	0.0004399	Paxs	508.27	Joback Method
dvisc	0.0010327	Paxs	438.53	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U392442&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U392442&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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