

# Fumaric acid, octyl 4-phenylphenyl ester

<b>Inchi:</b>	InChI=1S/C24H28O4/c1-2-3-4-5-6-10-19-27-23(25)17-18-24(26)28-22-15-13-21(14-16-2
<b>InchiKey:</b>	QLZHLASOKWEOFJ-ISLYRVAYSA-N
<b>Formula:</b>	C24H28O4
<b>SMILES:</b>	CCCCCCCCOC(=O)C=CC(=O)Oc1ccc(-c2ccccc2)cc1
<b>Mol. weight [g/mol]:</b>	380.48

## Physical Properties

Property code	Value	Unit	Source
gf	-21.23	kJ/mol	Joback Method
hf	-449.48	kJ/mol	Joback Method
hfus	51.38	kJ/mol	Joback Method
hvap	92.50	kJ/mol	Joback Method
log10ws	-7.29		Crippen Method
logp	5.719		Crippen Method
mvol	312.080	ml/mol	McGowan Method
pc	1340.78	kPa	Joback Method
rinpol	3176.00		NIST Webbook
rinpol	3176.00		NIST Webbook
tb	963.60	K	Joback Method
tc	1190.00	K	Joback Method
tf	564.84	K	Joback Method
vc	1.192	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	988.12	J/molxK	963.60	Joback Method
cpg	1002.36	J/molxK	1001.33	Joback Method
cpg	1015.35	J/molxK	1039.07	Joback Method
cpg	1027.15	J/molxK	1076.80	Joback Method
cpg	1037.85	J/molxK	1114.53	Joback Method
cpg	1047.50	J/molxK	1152.26	Joback Method
cpg	1056.19	J/molxK	1190.00	Joback Method
dvisc	0.0003126	Paxs	564.84	Joback Method

dvisc	0.0001690	Paxs	631.30	Joback Method
dvisc	0.0001028	Paxs	697.76	Joback Method
dvisc	0.0000681	Paxs	764.22	Joback Method
dvisc	0.0000482	Paxs	830.68	Joback Method
dvisc	0.0000359	Paxs	897.14	Joback Method
dvisc	0.0000279	Paxs	963.60	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348214&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348214&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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