

# Glutaric acid, hept-2-yl trans-4-tert-butylcyclohexyl ester

<b>Inchi:</b>	InChI=1S/C22H40O4/c1-6-7-8-10-17(2)25-20(23)11-9-12-21(24)26-19-15-13-18(14-16-1
<b>InchiKey:</b>	BYBZFAUTJXZQBV-UHFFFAOYSA-N
<b>Formula:</b>	C22H40O4
<b>SMILES:</b>	CCCCC(C)OC(=O)CCCC(=O)OC1CCC(C(C)(C)C)CC1
<b>Mol. weight [g/mol]:</b>	368.55

## Physical Properties

Property code	Value	Unit	Source
gf	-316.34	kJ/mol	Joback Method
hf	-967.06	kJ/mol	Joback Method
hfus	40.28	kJ/mol	Joback Method
hvap	81.31	kJ/mol	Joback Method
log10ws	-6.39		Crippen Method
logp	5.817		Crippen Method
mvol	324.860	ml/mol	McGowan Method
pc	1082.78	kPa	Joback Method
rinpol	2475.00		NIST Webbook
rinpol	2475.00		NIST Webbook
tb	866.55	K	Joback Method
tc	1069.84	K	Joback Method
tf	472.58	K	Joback Method
vc	1.230	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1093.84	J/molxK	866.55	Joback Method
cpg	1113.55	J/molxK	900.43	Joback Method
cpg	1131.78	J/molxK	934.31	Joback Method
cpg	1148.59	J/molxK	968.19	Joback Method
cpg	1164.01	J/molxK	1002.07	Joback Method
cpg	1178.09	J/molxK	1035.96	Joback Method
cpg	1190.86	J/molxK	1069.84	Joback Method
dvisc	0.0008924	Paxs	472.58	Joback Method

dvisc	0.0003833	Paxs	538.24	Joback Method
dvisc	0.0001979	Paxs	603.90	Joback Method
dvisc	0.0001163	Paxs	669.56	Joback Method
dvisc	0.0000751	Paxs	735.23	Joback Method
dvisc	0.0000522	Paxs	800.89	Joback Method
dvisc	0.0000383	Paxs	866.55	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U393403&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U393403&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/84-295-2/Glutaric-acid-hept-2-yl-trans-4-tert-butylcyclohexyl-ester.pdf>

Generated by Cheméo on 2024-04-26 08:28:36.854987059 +0000 UTC m=+16409365.775564381.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.