

# Oligandrol

<b>Inchi:</b>	InChI=1S/C22H32O2/c1-16(2)8-6-9-17(3)10-7-12-22(5)13-11-19-15-20(23)14-18(4)21(19)
<b>InchiKey:</b>	OJZATKPMGAGZGS-LICLKQGHSA-N
<b>Formula:</b>	C22H32O2
<b>SMILES:</b>	CC(C)=CCCC(C)=CCCC1(C)CCc2cc(O)cc(C)c2O1
<b>Mol. weight [g/mol]:</b>	328.49
<b>CAS:</b>	155661-15-7

## Physical Properties

Property code	Value	Unit	Source
gf	173.27	kJ/mol	Joback Method
hf	-296.39	kJ/mol	Joback Method
hfus	47.28	kJ/mol	Joback Method
hvap	84.70	kJ/mol	Joback Method
log10ws	-7.15		Crippen Method
logp	6.257		Crippen Method
mcvol	289.360	ml/mol	McGowan Method
pc	1508.15	kPa	Joback Method
rinpol	2710.30		NIST Webbook
rinpol	2710.30		NIST Webbook
tb	866.30	K	Joback Method
tc	1092.99	K	Joback Method
tf	527.69	K	Joback Method
vc	1.056	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	922.89	J/molxK	866.30	Joback Method
cpg	944.09	J/molxK	904.08	Joback Method
cpg	965.35	J/molxK	941.86	Joback Method
cpg	986.94	J/molxK	979.65	Joback Method
cpg	1009.13	J/molxK	1017.43	Joback Method
cpg	1032.18	J/molxK	1055.21	Joback Method
cpg	1056.35	J/molxK	1092.99	Joback Method

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C155661157&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C155661157&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>r in pol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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