

# 3«alpha»-Propanoyloxy-5«alpha»-androstan-17-o

<b>Other names:</b>	Etiocholan-3«alpha»-ol-17-one, propionate
<b>Inchi:</b>	InChI=1S/C22H34O3/c1-4-20(24)25-15-9-11-21(2)14(13-15)5-6-16-17-7-8-19(23)22(17,3
<b>InchiKey:</b>	SAOVUTZAPBDLOX-VYQAZBKGSA-N
<b>Formula:</b>	C22H34O3
<b>SMILES:</b>	CCC(=O)OC1CCC2(C)C(CCC3C4CCC(=O)C4(C)CCC32)C1
<b>Mol. weight [g/mol]:</b>	346.50
<b>CAS:</b>	5953-68-4

## Physical Properties

Property code	Value	Unit	Source
gf	-73.76	kJ/mol	Joback Method
hf	-650.05	kJ/mol	Joback Method
hfus	27.69	kJ/mol	Joback Method
hvap	75.25	kJ/mol	Joback Method
log10ws	-5.41		Crippen Method
logp	4.920		Crippen Method
mcvol	286.410	ml/mol	McGowan Method
pc	1464.61	kPa	Joback Method
rinpol	2400.00		NIST Webbook
tb	881.65	K	Joback Method
tc	1123.30	K	Joback Method
tf	567.32	K	Joback Method
vc	1.079	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1041.52	J/molxK	881.65	Joback Method
cpg	1069.44	J/molxK	921.93	Joback Method
cpg	1097.22	J/molxK	962.20	Joback Method
cpg	1125.19	J/molxK	1002.48	Joback Method
cpg	1153.67	J/molxK	1042.75	Joback Method
cpg	1182.99	J/molxK	1083.03	Joback Method
cpg	1213.47	J/molxK	1123.30	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5953684&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5953684&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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