

# cis-Cyclohex-4-en-1,2-dicarboxylic acid, 2-ethylhexyl isoheptyl ester

Inchi:	InChI=1S/C22H38O4/c1-5-7-12-18(6-2)16-26-22(24)20-14-9-8-13-19(20)21(23)25-15-10
InchiKey:	FVCVEJYVDPWSDB-UHFFFAOYSA-N
Formula:	C22H38O4
SMILES:	CCCCC(CC)COC(=O)C1CC=CCC1C(=O)OCCCC(C)C
Mol. weight [g/mol]:	366.53

## Physical Properties

Property code	Value	Unit	Source
gf	-291.66	kJ/mol	Joback Method
hf	-905.81	kJ/mol	Joback Method
hfus	45.39	kJ/mol	Joback Method
hvap	82.51	kJ/mol	Joback Method
log10ws	-5.54		Crippen Method
logp	5.308		Crippen Method
mcvol	320.560	ml/mol	McGowan Method
pc	1100.81	kPa	Joback Method
rinpol	2360.00		NIST Webbook
tb	868.50	K	Joback Method
tc	1070.02	K	Joback Method
tf	455.92	K	Joback Method
vc	1.222	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1064.89	J/molxK	868.50	Joback Method
cpg	1083.83	J/molxK	902.09	Joback Method
cpg	1101.33	J/molxK	935.67	Joback Method
cpg	1117.40	J/molxK	969.26	Joback Method
cpg	1132.07	J/molxK	1002.85	Joback Method
cpg	1145.36	J/molxK	1036.44	Joback Method
cpg	1157.31	J/molxK	1070.02	Joback Method
dvisc	0.0010737	Paxs	455.92	Joback Method
dvisc	0.0004565	Paxs	524.68	Joback Method

dvisc	0.0002367	Paxs	593.45	Joback Method
dvisc	0.0001406	Paxs	662.21	Joback Method
dvisc	0.0000922	Paxs	730.97	Joback Method
dvisc	0.0000649	Paxs	799.74	Joback Method
dvisc	0.0000484	Paxs	868.50	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U382632&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U382632&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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