

# Isophthalic acid, 2-biphenyl ethyl ester

<b>Inchi:</b>	InChI=1S/C22H18O4/c1-2-25-21(23)17-11-8-12-18(15-17)22(24)26-20-14-7-6-13-19(20)
<b>InchiKey:</b>	OXWBJNLVOLBYDT-UHFFFAOYSA-N
<b>Formula:</b>	C22H18O4
<b>SMILES:</b>	CCOC(=O)c1cccc(C(=O)Oc2ccccc2-c2ccccc2)c1
<b>Mol. weight [g/mol]:</b>	346.38

## Physical Properties

Property code	Value	Unit	Source
gf	-15.51	kJ/mol	Joback Method
hf	-300.36	kJ/mol	Joback Method
hfus	39.65	kJ/mol	Joback Method
hvap	91.03	kJ/mol	Joback Method
log10ws	-6.82		Crippen Method
logp	4.749		Crippen Method
mvol	264.440	ml/mol	McGowan Method
pc	1932.13	kPa	Joback Method
rinpol	2803.00		NIST Webbook
rinpol	2803.00		NIST Webbook
tb	945.34	K	Joback Method
tc	1195.68	K	Joback Method
tf	586.32	K	Joback Method
vc	0.992	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	792.30	J/molxK	945.34	Joback Method
cpg	837.97	J/molxK	1153.96	Joback Method
cpg	831.57	J/molxK	1112.24	Joback Method
cpg	823.87	J/molxK	1070.51	Joback Method
cpg	814.80	J/molxK	1028.79	Joback Method
cpg	804.30	J/molxK	987.06	Joback Method
cpg	843.14	J/molxK	1195.68	Joback Method
dvisc	0.0000448	Paxs	945.34	Joback Method

dvisc	0.0000559	Paxs	885.50	Joback Method
dvisc	0.0000722	Paxs	825.67	Joback Method
dvisc	0.0000969	Paxs	765.83	Joback Method
dvisc	0.0001369	Paxs	705.99	Joback Method
dvisc	0.0002060	Paxs	646.16	Joback Method
dvisc	0.0003370	Paxs	586.32	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U344556&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U344556&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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