

Zearalenone bis(trifluoroacetate)

Other names:	17924-92-4, F3Ac
Inchi:	InChI=1S/C22H20F6O7/c1-12-6-5-9-14(29)8-4-2-3-7-13-10-15(34-19(31)21(23,24)25)11
InchiKey:	HPXZUGAUANXHJR-XVNBXDOJSA-N
Formula:	C22H20F6O7
SMILES:	CC1CCCC(=O)CCCC=Cc2cc(OC(=O)C(F)(F)F)cc(OC(=O)C(F)(F)F)c2C(=O)O1
Mol. weight [g/mol]:	510.38

Physical Properties

Property code	Value	Unit	Source
gf	-1762.63	kJ/mol	Joback Method
hf	-2311.31	kJ/mol	Joback Method
hfus	42.29	kJ/mol	Joback Method
hvap	94.40	kJ/mol	Joback Method
log10ws	-7.13		Crippen Method
logp	5.104		Crippen Method
mcvol	316.430	ml/mol	McGowan Method
pc	1339.80	kPa	Joback Method
rinpol	2409.00		NIST Webbook
rinpol	2409.00		NIST Webbook
tb	1093.04	K	Joback Method
tc	1342.30	K	Joback Method
tf	704.41	K	Joback Method
vc	1.200	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1084.52	J/molxK	1093.04	Joback Method
cpg	1085.97	J/molxK	1134.58	Joback Method
cpg	1084.03	J/molxK	1176.13	Joback Method
cpg	1078.64	J/molxK	1217.67	Joback Method
cpg	1069.74	J/molxK	1259.21	Joback Method
cpg	1057.25	J/molxK	1300.75	Joback Method
cpg	1041.12	J/molxK	1342.30	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U375714&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvpap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinppl:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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