

# Phthalic acid, hexyl 2-methoxybenzyl ester

<b>Inchi:</b>	InChI=1S/C22H26O5/c1-3-4-5-10-15-26-21(23)18-12-7-8-13-19(18)22(24)27-16-17-11-6
<b>InchiKey:</b>	ZAOSZPSCZFQDGP-UHFFFAOYSA-N
<b>Formula:</b>	C22H26O5
<b>SMILES:</b>	CCCCCOC(=O)c1ccccc1C(=O)OCc1ccccc1OC
<b>Mol. weight [g/mol]:</b>	370.44

## Physical Properties

Property code	Value	Unit	Source
gf	-232.92	kJ/mol	Joback Method
hf	-669.11	kJ/mol	Joback Method
hfus	46.80	kJ/mol	Joback Method
hvap	91.16	kJ/mol	Joback Method
log10ws	-6.28		Crippen Method
logp	4.789		Crippen Method
mcvol	294.070	ml/mol	McGowan Method
pc	1453.46	kPa	Joback Method
rinpol	3167.00		NIST Webbook
rinpol	3167.00		NIST Webbook
tb	941.08	K	Joback Method
tc	1164.13	K	Joback Method
tf	582.13	K	Joback Method
vc	1.117	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	924.20	J/molxK	941.08	Joback Method
cpg	937.58	J/molxK	978.26	Joback Method
cpg	949.51	J/molxK	1015.43	Joback Method
cpg	960.00	J/molxK	1052.61	Joback Method
cpg	969.08	J/molxK	1089.78	Joback Method
cpg	976.76	J/molxK	1126.96	Joback Method
cpg	983.08	J/molxK	1164.13	Joback Method
dvisc	0.0002639	Paxs	582.13	Joback Method

dvisc	0.0001580	Paxs	641.95	Joback Method
dvisc	0.0001032	Paxs	701.78	Joback Method
dvisc	0.0000721	Paxs	761.61	Joback Method
dvisc	0.0000531	Paxs	821.43	Joback Method
dvisc	0.0000407	Paxs	881.25	Joback Method
dvisc	0.0000323	Paxs	941.08	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U382495&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U382495&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/84-607-5/Phthalic-acid-hexyl-2-methoxybenzyl-ester.pdf>

Generated by Cheméo on 2024-09-08 06:49:51.426145193 +0000 UTC m=+359054.063114441.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.