

# Isophthalic acid, 4-formylphenyl hexyl ester

<b>Inchi:</b>	InChI=1S/C21H22O5/c1-2-3-4-5-13-25-20(23)17-7-6-8-18(14-17)21(24)26-19-11-9-16(15)
<b>InchiKey:</b>	RAHJVMVBTGGAGP-UHFFFAOYSA-N
<b>Formula:</b>	C21H22O5
<b>SMILES:</b>	CCCCCCOC(=O)c1cccc(C(=O)Oc2ccc(C=O)cc2)c1
<b>Mol. weight [g/mol]:</b>	354.40

## Physical Properties

Property code	Value	Unit	Source
gf	-235.86	kJ/mol	Joback Method
hf	-601.83	kJ/mol	Joback Method
hfus	45.31	kJ/mol	Joback Method
hvap	93.25	kJ/mol	Joback Method
log10ws	-6.14		Crippen Method
logp	4.455		Crippen Method
mvol	275.680	ml/mol	McGowan Method
pc	1685.18	kPa	Joback Method
rinpol	3030.00		NIST Webbook
rinpol	3030.00		NIST Webbook
tb	944.44	K	Joback Method
tc	1171.66	K	Joback Method
tf	590.63	K	Joback Method
vc	1.060	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	848.27	J/molxK	944.44	Joback Method
cpg	860.27	J/molxK	982.31	Joback Method
cpg	870.96	J/molxK	1020.18	Joback Method
cpg	880.38	J/molxK	1058.05	Joback Method
cpg	888.57	J/molxK	1095.92	Joback Method
cpg	895.55	J/molxK	1133.79	Joback Method
cpg	901.37	J/molxK	1171.66	Joback Method
dvisc	0.0004047	Paxs	590.63	Joback Method

dvisc	0.0002482	Paxs	649.60	Joback Method
dvisc	0.0001651	Paxs	708.57	Joback Method
dvisc	0.0001170	Paxs	767.53	Joback Method
dvisc	0.0000870	Paxs	826.50	Joback Method
dvisc	0.0000673	Paxs	885.47	Joback Method
dvisc	0.0000538	Paxs	944.44	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U344695&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U344695&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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