

# 3,5-Heptanedione, 1,7-bis(4-hydroxy-3-methoxyphenyl)-

<b>Other names:</b>	Tetrahydrodiferuloylmethane Tetrahydrocurcumin 1,7-bis(4-Hydroxy-3-methoxyphenyl)heptane-3,5-dione
<b>Inchi:</b>	InChI=1S/C21H24O6/c1-26-20-11-14(5-9-18(20)24)3-7-16(22)13-17(23)8-4-15-6-10-19(2)
<b>InchiKey:</b>	LBTVHXHERHESKG-UHFFFAOYSA-N
<b>Formula:</b>	C21H24O6
<b>SMILES:</b>	COc1cc(CCC(=O)CC(=O)CCc2ccc(O)c(OC)c2)ccc1O
<b>Mol. weight [g/mol]:</b>	372.41
<b>CAS:</b>	36062-04-1

## Physical Properties

Property code	Value	Unit	Source
gf	-445.58	kJ/mol	Joback Method
hf	-870.87	kJ/mol	Joback Method
hfus	54.59	kJ/mol	Joback Method
hvap	112.56	kJ/mol	Joback Method
log10ws	-3.88		Crippen Method
logp	3.209		Crippen Method
mcvol	285.850	ml/mol	McGowan Method
pc	2043.76	kPa	Joback Method
rinsol	3187.60		NIST Webbook
rinsol	3187.60		NIST Webbook
tb	1057.02	K	Joback Method
tc	1301.28	K	Joback Method
tf	772.07	K	Joback Method
vc	0.976	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	940.89	J/mol×K	1057.02	Joback Method
cpg	955.75	J/mol×K	1097.73	Joback Method
cpg	970.44	J/mol×K	1138.44	Joback Method
cpg	985.09	J/mol×K	1179.15	Joback Method

cpg	999.88	J/molxK	1219.86	Joback Method
cpg	1014.94	J/molxK	1260.57	Joback Method
cpg	1030.44	J/molxK	1301.28	Joback Method
dvisc	0.0000007	Paxs	772.07	Joback Method
dvisc	0.0000004	Paxs	819.56	Joback Method
dvisc	0.0000002	Paxs	867.05	Joback Method
dvisc	0.0000001	Paxs	914.54	Joback Method
dvisc	8.9906777e-08	Paxs	962.04	Joback Method
dvisc	6.0756764e-08	Paxs	1009.53	Joback Method
dvisc	4.2529545e-08	Paxs	1057.02	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C36062041&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C36062041&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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