

# Succinic acid, 4-chloro-3-methylphenyl dec-4-en-1-yl ester

<b>Inchi:</b>	InChI=1S/C21H29ClO4/c1-3-4-5-6-7-8-9-10-15-25-20(23)13-14-21(24)26-18-11-12-19(22)
<b>InchiKey:</b>	DHGLFRHRFNOHFR-BQYQJAHWSA-N
<b>Formula:</b>	C21H29ClO4
<b>SMILES:</b>	CCCCC=CCCCOC(=O)CCC(=O)Oc1ccc(Cl)c(C)c1
<b>Mol. weight [g/mol]:</b>	380.91

## Physical Properties

Property code	Value	Unit	Source
gf	-180.46	kJ/mol	Joback Method
hf	-651.30	kJ/mol	Joback Method
hfus	53.38	kJ/mol	Joback Method
hvap	88.60	kJ/mol	Joback Method
log10ws	-6.69		Crippen Method
logp	5.794		Crippen Method
mvol	305.810	ml/mol	McGowan Method
pc	1252.15	kPa	Joback Method
rinpol	2791.00		NIST Webbook
rinpol	2791.00		NIST Webbook
tb	910.69	K	Joback Method
tc	1121.82	K	Joback Method
tf	547.05	K	Joback Method
vc	1.181	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	935.73	J/molxK	910.69	Joback Method
cpg	950.43	J/molxK	945.88	Joback Method
cpg	964.01	J/molxK	981.07	Joback Method
cpg	976.50	J/molxK	1016.26	Joback Method
cpg	987.93	J/molxK	1051.45	Joback Method
cpg	998.36	J/molxK	1086.64	Joback Method
cpg	1007.80	J/molxK	1121.82	Joback Method
dvisc	0.0003504	Paxs	547.05	Joback Method

dvisc	0.0001990	Paxs	607.66	Joback Method
dvisc	0.0001252	Paxs	668.26	Joback Method
dvisc	0.0000851	Paxs	728.87	Joback Method
dvisc	0.0000614	Paxs	789.48	Joback Method
dvisc	0.0000464	Paxs	850.08	Joback Method
dvisc	0.0000364	Paxs	910.69	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U391179&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U391179&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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