

dvisc	0.0000150	Paxs	844.93	Joback Method
dvisc	0.0000196	Paxs	789.01	Joback Method
dvisc	0.0000266	Paxs	733.10	Joback Method
dvisc	0.0000382	Paxs	677.18	Joback Method
dvisc	0.0000585	Paxs	621.26	Joback Method
dvisc	0.0000973	Paxs	565.35	Joback Method
dvisc	0.0001811	Paxs	509.43	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R188191&Units=SI

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/84-823-5/2-2-2-2-Pentoxo-ethoxy-ethoxy-ethoxy-ethoxy-ethyl-acetate.pdf>

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