

Hexanamide, 6-chloro-N-ethyl-N-heptyl-

Inchi:	InChI=1S/C15H30ClNO/c1-3-5-6-7-11-14-17(4-2)15(18)12-9-8-10-13-16/h3-14H2,1-2H3
InchiKey:	MFYDFNQNIFODJG-UHFFFAOYSA-N
Formula:	C15H30ClNO
SMILES:	CCCCCCCN(CC)C(=O)CCCCCl
Mol. weight [g/mol]:	275.86

Physical Properties

Property code	Value	Unit	Source
gf	45.35	kJ/mol	Joback Method
hf	-413.72	kJ/mol	Joback Method
hfus	43.42	kJ/mol	Joback Method
hvap	62.16	kJ/mol	Joback Method
log10ws	-4.60		Crippen Method
logp	4.605		Crippen Method
mvol	246.000	ml/mol	McGowan Method
pc	1441.35	kPa	Joback Method
rinpol	2275.00		NIST Webbook
rinpol	2275.00		NIST Webbook
tb	646.34	K	Joback Method
tc	817.60	K	Joback Method
tf	371.13	K	Joback Method
vc	0.949	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	665.27	J/mol×K	646.34	Joback Method
cpg	682.45	J/mol×K	674.88	Joback Method
cpg	698.83	J/mol×K	703.43	Joback Method
cpg	714.43	J/mol×K	731.97	Joback Method
cpg	729.29	J/mol×K	760.52	Joback Method
cpg	743.43	J/mol×K	789.06	Joback Method
cpg	756.89	J/mol×K	817.60	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U415594&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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