

# Methyl benzilate

**Other names:**

Benzeneacetic acid, «alpha»-hydroxy-«alpha»-phenyl-, methyl ester  
Benzilic acid, methyl ester  
Methyl diphenylglycolate  
Methyl «alpha»-hydroxydiphenylacetate  
Methyl «alpha»-phenylmandelate  
Methyl 2-hydroxy-2,2-diphenylacetate  
Methyl hydroxydiphenylacetate  
Hydroxy-diphenyl-acetic acid methyl ester  
«alpha»-Hydroxydiphenylacetic acid, methyl ester  
NSC 57672

**Inchi:**

InChI=1S/C15H14O3/c1-18-14(16)15(17,12-8-4-2-5-9-12)13-10-6-3-7-11-13/h2-11,17H,1

**InchiKey:**

LJFIHTFNTGQZJL-UHFFFAOYSA-N

**Formula:**

C15H14O3

**SMILES:**

COC(=O)C(O)(c1ccccc1)c1ccccc1

**Mol. weight [g/mol]:**

242.27

**CAS:**

76-89-1

## Physical Properties

Property code	Value	Unit	Source
gf	-67.66	kJ/mol	Joback Method
hf	-285.65	kJ/mol	Joback Method
hfus	22.15	kJ/mol	Joback Method
hvap	78.08	kJ/mol	Joback Method
log10ws	-2.70		Crippen Method
logp	2.095		Crippen Method
mcvol	188.000	ml/mol	McGowan Method
pc	2956.90	kPa	Joback Method
rinpol	1805.00		NIST Webbook
rinpol	1792.00		NIST Webbook
rinpol	1825.00		NIST Webbook
rinpol	1853.00		NIST Webbook
rinpol	1853.00		NIST Webbook
ripol	2833.00		NIST Webbook
ripol	2852.00		NIST Webbook
tb	761.20	K	Joback Method
tc	991.88	K	Joback Method
tf	347.50 ± 0.50	K	NIST Webbook

vc

0.692

m<sup>3</sup>/kmol

Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	521.81	J/mol×K	761.20	Joback Method
cpg	534.28	J/mol×K	799.65	Joback Method
cpg	545.67	J/mol×K	838.09	Joback Method
cpg	556.08	J/mol×K	876.54	Joback Method
cpg	565.59	J/mol×K	914.99	Joback Method
cpg	574.27	J/mol×K	953.43	Joback Method
cpg	582.21	J/mol×K	991.88	Joback Method
dvisc	0.0010065	Paxs	447.05	Joback Method
dvisc	0.0003832	Paxs	499.41	Joback Method
dvisc	0.0001752	Paxs	551.77	Joback Method
dvisc	0.0000918	Paxs	604.12	Joback Method
dvisc	0.0000533	Paxs	656.48	Joback Method
dvisc	0.0000335	Paxs	708.84	Joback Method
dvisc	0.0000225	Paxs	761.20	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	460.20	K	1.70	NIST Webbook
tbrp	460.00	K	1.70	NIST Webbook

## Sources

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C76891&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/84-898-3/Methyl-benzilate.pdf>

Generated by Cheméo on 2024-04-27 05:43:50.97218992 +0000 UTC m=+16485879.892767236.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.