

# Fumaric acid, 3-methylbutyl 3-chlorophenyl ester

Inchi:	InChI=1S/C15H17ClO4/c1-11(2)8-9-19-14(17)6-7-15(18)20-13-5-3-4-12(16)10-13/h3-7,1
InchiKey:	PBMDALNQYYTTLJ-VOTSOKGWSA-N
Formula:	C15H17ClO4
SMILES:	CC(C)CCOC(=O)C=CC(=O)Oc1cccc(Cl)c1
Mol. weight [g/mol]:	296.75

## Physical Properties

Property code	Value	Unit	Source
gf	-223.79	kJ/mol	Joback Method
hf	-521.27	kJ/mol	Joback Method
hfus	34.71	kJ/mol	Joback Method
hvap	74.19	kJ/mol	Joback Method
log10ws	-3.87		Crippen Method
logp	3.391		Crippen Method
mcvol	221.270	ml/mol	McGowan Method
pc	2016.32	kPa	Joback Method
rinpol	2073.00		NIST Webbook
rinpol	2073.00		NIST Webbook
tb	767.99	K	Joback Method
tc	985.06	K	Joback Method
tf	451.91	K	Joback Method
vc	0.839	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	595.55	J/molxK	767.99	Joback Method
cpg	608.89	J/molxK	804.17	Joback Method
cpg	621.26	J/molxK	840.35	Joback Method
cpg	632.68	J/molxK	876.53	Joback Method
cpg	643.18	J/molxK	912.71	Joback Method
cpg	652.79	J/molxK	948.89	Joback Method
cpg	661.53	J/molxK	985.06	Joback Method
dvisc	0.0008258	Paxs	451.91	Joback Method

dvisc	0.0004501	Paxs	504.59	Joback Method
dvisc	0.0002752	Paxs	557.27	Joback Method
dvisc	0.0001831	Paxs	609.95	Joback Method
dvisc	0.0001300	Paxs	662.63	Joback Method
dvisc	0.0000971	Paxs	715.31	Joback Method
dvisc	0.0000755	Paxs	767.99	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405552&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405552&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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