

# Glutaric acid, 2-methylpent-3-yl but-3-en-1-yl ester

Inchi:	InChI=1S/C15H26O4/c1-5-7-11-18-14(16)9-8-10-15(17)19-13(6-2)12(3)4/h5,12-13H,1,6-
InchiKey:	NPAXMPOJQBPKGM-UHFFFAOYSA-N
Formula:	C15H26O4
SMILES:	C=CCCOC(=O)CCCC(=O)OC(CC)C(C)C
Mol. weight [g/mol]:	270.36

## Physical Properties

Property code	Value	Unit	Source
gf	-309.46	kJ/mol	Joback Method
hf	-727.66	kJ/mol	Joback Method
hfus	31.85	kJ/mol	Joback Method
hvap	65.85	kJ/mol	Joback Method
log10ws	-3.55		Crippen Method
logp	3.254		Crippen Method
mcvol	232.790	ml/mol	McGowan Method
pc	1589.81	kPa	Joback Method
rinpol	1702.00		NIST Webbook
rinpol	1702.00		NIST Webbook
tb	690.98	K	Joback Method
tc	873.40	K	Joback Method
tf	371.37	K	Joback Method
vc	0.892	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	654.33	J/molxK	690.98	Joback Method
cpg	670.39	J/molxK	721.38	Joback Method
cpg	685.64	J/molxK	751.79	Joback Method
cpg	700.07	J/molxK	782.19	Joback Method
cpg	713.72	J/molxK	812.59	Joback Method
cpg	726.57	J/molxK	842.99	Joback Method
cpg	738.65	J/molxK	873.40	Joback Method
dvisc	0.0020674	Paxs	371.37	Joback Method

dvisc	0.0008976	Paxs	424.64	Joback Method
dvisc	0.0004694	Paxs	477.91	Joback Method
dvisc	0.0002795	Paxs	531.17	Joback Method
dvisc	0.0001830	Paxs	584.44	Joback Method
dvisc	0.0001286	Paxs	637.71	Joback Method
dvisc	0.0000954	Paxs	690.98	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U394032&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U394032&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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