

# Dehydrocostus lactone

<b>Other names:</b>	(3aS,6aR,9aR,9bS)-3,6,9-Trimethylenedecahydroazuleno[4,5-b]furan-2(9bH)-one
<b>Inchi:</b>	InChI=1S/C15H18O2/c1-8-4-7-12-10(3)15(16)17-14(12)13-9(2)5-6-11(8)13/h11-14H,1-7
<b>InchiKey:</b>	NETSQGRTUNRXEO-UHFFFAOYSA-N
<b>Formula:</b>	C15H18O2
<b>SMILES:</b>	<chem>C=C1C(=O)OC2C1CCC(=C)C1CCC(=C)C12</chem>
<b>Mol. weight [g/mol]:</b>	230.30
<b>CAS:</b>	477-43-0

## Physical Properties

Property code	Value	Unit	Source
gf	152.09	kJ/mol	Joback Method
hf	-196.49	kJ/mol	Joback Method
hfus	25.69	kJ/mol	Joback Method
hvap	58.34	kJ/mol	Joback Method
log10ws	-3.60		Crippen Method
logp	3.017		Crippen Method
mcvol	184.170	ml/mol	McGowan Method
pc	2229.20	kPa	Joback Method
rinpol	2006.70		NIST Webbook
tb	667.48	K	Joback Method
tc	903.06	K	Joback Method
tf	430.14	K	Joback Method
vc	0.694	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	538.26	J/molxK	667.48	Joback Method
cpg	558.54	J/molxK	706.74	Joback Method
cpg	577.44	J/molxK	746.01	Joback Method
cpg	595.00	J/molxK	785.27	Joback Method
cpg	611.26	J/molxK	824.53	Joback Method
cpg	626.27	J/molxK	863.80	Joback Method
cpg	640.05	J/molxK	903.06	Joback Method

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C477430&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C477430&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>r in pol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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