

# «gamma»-Terpinyl acetate

<b>Other names:</b>	«gamma»-Terpenyl acetate 1-methyl-4-(1-methylethylidene)cyclohexyl acetate
<b>Inchi:</b>	InChI=1S/C12H20O2/c1-9(2)11-5-7-12(4,8-6-11)14-10(3)13/h5-8H2,1-4H3
<b>InchiKey:</b>	ZKKBZSOYCMSYRW-UHFFFAOYSA-N
<b>Formula:</b>	C12H20O2
<b>SMILES:</b>	CC(=O)OC1(C)CCC(=C(C)C)CC1
<b>Mol. weight [g/mol]:</b>	196.29
<b>CAS:</b>	10235-63-9

## Physical Properties

Property code	Value	Unit	Source
gf	-127.89	kJ/mol	Joback Method
hf	-400.01	kJ/mol	Joback Method
hfus	14.17	kJ/mol	Joback Method
hvap	51.61	kJ/mol	Joback Method
log10ws	-3.57		Crippen Method
logp	3.219		Crippen Method
mcvol	172.220	ml/mol	McGowan Method
pc	2402.92	kPa	Joback Method
rinpol	1336.00		NIST Webbook
rinpol	1256.40		NIST Webbook
rinpol	1341.30		NIST Webbook
rinpol	1354.00		NIST Webbook
ripol	1696.00		NIST Webbook
ripol	1696.00		NIST Webbook
ripol	1696.00		NIST Webbook
tb	576.56	K	Joback Method
tc	794.00	K	Joback Method
tf	324.84	K	Joback Method
vc	0.646	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	433.75	J/mol×K	576.56	Joback Method
cpg	451.92	J/mol×K	612.80	Joback Method
cpg	469.07	J/mol×K	649.04	Joback Method
cpg	485.31	J/mol×K	685.28	Joback Method
cpg	500.74	J/mol×K	721.52	Joback Method
cpg	515.48	J/mol×K	757.76	Joback Method
cpg	529.62	J/mol×K	794.00	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10235639&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10235639&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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