

# Diglycolic acid, butyl 2-formylphenyl ester

<b>Inchi:</b>	InChI=1S/C15H18O6/c1-2-3-8-20-14(17)10-19-11-15(18)21-13-7-5-4-6-12(13)9-16/h4-7,
<b>InchiKey:</b>	XGALRCKVSWZNI-UHFFFAOYSA-N
<b>Formula:</b>	C15H18O6
<b>SMILES:</b>	CCCCOC(=O)COCC(=O)Oc1ccccc1C=O
<b>Mol. weight [g/mol]:</b>	294.30

## Physical Properties

Property code	Value	Unit	Source
gf	-494.16	kJ/mol	Joback Method
hf	-835.27	kJ/mol	Joback Method
hfus	37.31	kJ/mol	Joback Method
hvap	79.36	kJ/mol	Joback Method
log10ws	-2.49		Crippen Method
logp	1.764		Crippen Method
mcvol	220.770	ml/mol	McGowan Method
pc	2079.33	kPa	Joback Method
rinpol	2706.00		NIST Webbook
rinpol	2706.00		NIST Webbook
tb	797.92	K	Joback Method
tc	1003.87	K	Joback Method
tf	506.30	K	Joback Method
vc	0.851	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	635.85	J/molxK	797.92	Joback Method
cpg	648.49	J/molxK	832.25	Joback Method
cpg	660.15	J/molxK	866.57	Joback Method
cpg	670.80	J/molxK	900.90	Joback Method
cpg	680.46	J/molxK	935.22	Joback Method
cpg	689.11	J/molxK	969.55	Joback Method
cpg	696.76	J/molxK	1003.87	Joback Method
dvisc	0.0006475	Paxs	506.30	Joback Method

dvisc	0.0004019	Paxs	554.90	Joback Method
dvisc	0.0002694	Paxs	603.51	Joback Method
dvisc	0.0001917	Paxs	652.11	Joback Method
dvisc	0.0001430	Paxs	700.71	Joback Method
dvisc	0.0001108	Paxs	749.32	Joback Method
dvisc	0.0000885	Paxs	797.92	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U382309&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U382309&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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