

# 2-Octyne, 1,1-diethoxy-

<b>Other names:</b>	1,1-diethoxyoct-2-yne
<b>Inchi:</b>	InChI=1S/C12H22O2/c1-4-7-8-9-10-11-12(13-5-2)14-6-3/h12H,4-9H2,1-3H3
<b>InchiKey:</b>	YKXAYRBNHIHDSI-UHFFFAOYSA-N
<b>Formula:</b>	C12H22O2
<b>SMILES:</b>	CCCCC#CC(OCC)OCC
<b>Mol. weight [g/mol]:</b>	198.30
<b>CAS:</b>	16387-55-6

## Physical Properties

Property code	Value	Unit	Source
chl	-7492.30	kJ/mol	NIST Webbook
gf	40.52	kJ/mol	Joback Method
hf	-288.43	kJ/mol	Joback Method
hfl	-374.00 ± 6.00	kJ/mol	NIST Webbook
hfl	443.10	kJ/mol	NIST Webbook
hfus	28.81	kJ/mol	Joback Method
hvap	48.89	kJ/mol	Joback Method
log10ws	-3.42		Crippen Method
logp	2.969		Crippen Method
mcvol	183.080	ml/mol	McGowan Method
pc	2000.12	kPa	Joback Method
tb	527.36	K	Joback Method
tc	710.36	K	Joback Method
tf	360.56	K	Joback Method
vc	0.700	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	431.15	J/mol×K	527.36	Joback Method
cpg	447.28	J/mol×K	557.86	Joback Method
cpg	462.82	J/mol×K	588.36	Joback Method
cpg	477.77	J/mol×K	618.86	Joback Method
cpg	492.13	J/mol×K	649.36	Joback Method

cpg	505.89	J/mol×K	679.86	Joback Method
cpg	519.07	J/mol×K	710.36	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C16387556&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C16387556&amp;Units=SI</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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