

# Levo-menthoxyacetic acid

<b>Other names:</b>	(-)-Menthyloxyacetic acid L-Menthoxyacetic acid (-)-Menthoxyacetic acid Acetic acid, [[5-methyl-2-(1-methylethyl)cyclohexyl]oxy]-, [1R-(1«alpha», 2«beta», 5«alpha»)]- L-p-menth-3-ylloxyacetic acid
<b>Inchi:</b>	InChI=1S/C12H22O3/c1-8(2)10-5-4-9(3)6-11(10)15-7-12(13)14/h8-11H,4-7H2,1-3H3,(H,
<b>InchiKey:</b>	CILPHQCEVYJUDN-UHFFFAOYSA-N
<b>Formula:</b>	C12H22O3
<b>SMILES:</b>	CC1CCC(C(C)C)C(OCC(=O)O)C1
<b>Mol. weight [g/mol]:</b>	214.30
<b>CAS:</b>	40248-63-3

## Physical Properties

Property code	Value	Unit	Source
gf	-313.99	kJ/mol	Joback Method
hf	-679.68	kJ/mol	Joback Method
hfus	24.17	kJ/mol	Joback Method
hvap	67.56	kJ/mol	Joback Method
log10ws	-2.31		Crippen Method
logp	2.548		Crippen Method
mcvol	182.390	ml/mol	McGowan Method
pc	2300.32	kPa	Joback Method
tb	652.20	K	Joback Method
tc	843.77	K	Joback Method
tf	341.88	K	Joback Method
vc	0.675	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	526.75	J/molxK	652.20	Joback Method
cpg	543.67	J/molxK	684.13	Joback Method
cpg	559.69	J/molxK	716.06	Joback Method
cpg	574.81	J/molxK	747.99	Joback Method

cpg	589.05	J/molxK	779.91	Joback Method
cpg	602.41	J/molxK	811.84	Joback Method
cpg	614.89	J/molxK	843.77	Joback Method
dvisc	0.0061433	Paxs	341.88	Joback Method
dvisc	0.0017993	Paxs	393.60	Joback Method
dvisc	0.0007009	Paxs	445.32	Joback Method
dvisc	0.0003322	Paxs	497.04	Joback Method
dvisc	0.0001813	Paxs	548.76	Joback Method
dvisc	0.0001098	Paxs	600.48	Joback Method
dvisc	0.0000720	Paxs	652.20	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	436.70	K	1.30	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C40248633&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C40248633&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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