

# 2-(2-(2-Isobutoxy-ethoxy)-ethoxy)-ethoxy)-ethanol

<b>Inchi:</b>	InChI=1S/C12H26O5/c1-12(2)11-17-10-9-16-8-7-15-6-5-14-4-3-13/h12-13H,3-11H2,1-2H
<b>InchiKey:</b>	DCNSLUGOHIBLOR-UHFFFAOYSA-N
<b>Formula:</b>	C12H26O5
<b>SMILES:</b>	CC(C)COCCOCCOCCOCCO
<b>Mol. weight [g/mol]:</b>	250.33

## Physical Properties

Property code	Value	Unit	Source
gf	-509.10	kJ/mol	Joback Method
hf	-977.40	kJ/mol	Joback Method
hfus	32.15	kJ/mol	Joback Method
hvap	68.24	kJ/mol	Joback Method
log10ws	-0.22		Crippen Method
logp	0.701		Crippen Method
mvol	209.290	ml/mol	McGowan Method
pc	1829.41	kPa	Joback Method
rinpol	1799.60		NIST Webbook
tb	655.38	K	Joback Method
tc	818.40	K	Joback Method
tf	359.74	K	Joback Method
vc	0.792	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	591.77	J/molxK	655.38	Joback Method
cpg	658.54	J/molxK	791.23	Joback Method
cpg	646.34	J/molxK	764.06	Joback Method
cpg	633.56	J/molxK	736.89	Joback Method
cpg	620.20	J/molxK	709.72	Joback Method
cpg	606.26	J/molxK	682.55	Joback Method
cpg	670.14	J/molxK	818.40	Joback Method
dvisc	0.0000226	Paxs	655.38	Joback Method
dvisc	0.0000357	Paxs	606.11	Joback Method

dvisc	0.0000611	Paxs	556.83	Joback Method
dvisc	0.0001163	Paxs	507.56	Joback Method
dvisc	0.0002538	Paxs	458.29	Joback Method
dvisc	0.0006688	Paxs	409.01	Joback Method
dvisc	0.0022984	Paxs	359.74	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R188453&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R188453&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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